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The reaction of 2-cyanoacetamides **1** with carbon suboxide **2** afforded 6-amino-4-hydroxy-2(1*H*)-pyridones **4**. Compounds **4** were also obtained by reaction of amidines **1** and 2,4,6-trichlorophenylmalonates **3**.

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Several research groups have long shown an interest in malonylheterocycles such as 4-hydroxy-2-pyrones and -2-pyridones, and 6-hydroxypyrimid-4-ones for their biological properties.

Among the different synthetic methods available, one of the most versatile in the preparation of these compounds provides for the use of an appropriate 1,3-binucleophile with a malonic acid derivative [1]. The preparation of 4-hydroxy-2(1*H*)-pyridones is also known by the reaction of enamines and azomethines with malonic acid derivatives [2]. Pursuing studies on the reactivities of carbon suboxide and other malonic acid derivatives [3], in the present work we describe the reactions of 2-cyanoacetamides, extremely versatile synthons in the formation of heterocyclic

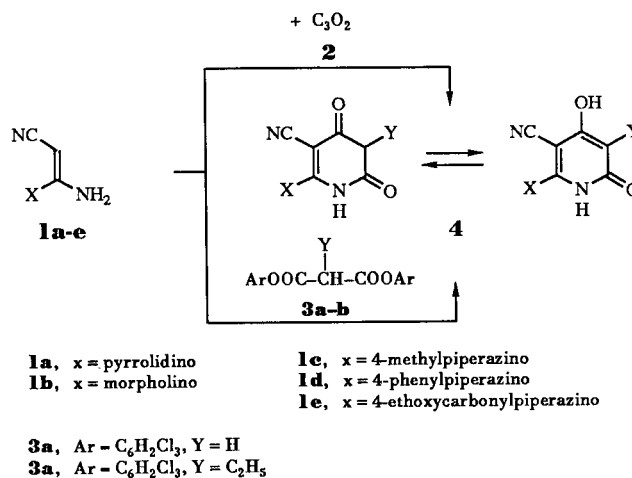
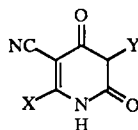


Table 1  
Physical and Analytical Data of Compounds **4**



Compound No.	X	Y	Method	Yield (%)	Mp (°C)	Formula	Analysis %		
							Calcd./	Found	
							C	H	N
<b>4a</b>	pyrrolidino	H	A	40	310 [a]	C <sub>10</sub> H <sub>11</sub> N <sub>3</sub> O <sub>2</sub>	58.53	5.40	20.48
				83			58.50	5.38	20.45
<b>4b</b>	morpholino	H	A	40	266 [b]	C <sub>10</sub> H <sub>11</sub> N <sub>3</sub> O <sub>3</sub>	54.23	5.01	19.00
				47			54.25	5.03	19.03
<b>4c</b>	4-methylpiperazino	H	A	53	210 [b]	C <sub>11</sub> H <sub>14</sub> N <sub>4</sub> O <sub>2</sub>	56.40	6.02	23.92
				85			56.45	6.00	23.89
<b>4d</b>	4-phenylpiperazino	H	A	40	255 [a]	C <sub>16</sub> H <sub>16</sub> N <sub>4</sub> O <sub>2</sub>	64.85	5.44	18.91
				84			64.80	5.42	18.93
<b>4e</b>	4-ethoxycarbonyl-piperazino	H	A	30	258 [c]	C <sub>13</sub> H <sub>16</sub> N <sub>4</sub> O <sub>4</sub>	53.42	5.52	19.17
				65			53.39	5.50	19.14
<b>4f</b>	pyrrolidino	C <sub>2</sub> H <sub>5</sub>	B	75	259 [a]	C <sub>12</sub> H <sub>15</sub> N <sub>3</sub> O <sub>2</sub>	61.78	6.48	18.02
							61.75	6.46	18.04
<b>4g</b>	morpholino	C <sub>2</sub> H <sub>5</sub>	B	46	265 [b]	C <sub>12</sub> H <sub>15</sub> N <sub>3</sub> O <sub>3</sub>	57.82	6.07	16.86
							57.79	6.05	16.84
<b>4h</b>	4-methylpiperazino	C <sub>2</sub> H <sub>5</sub>	B	60	265 [b]	C <sub>13</sub> H <sub>18</sub> N <sub>4</sub> O <sub>2</sub>	59.52	6.92	21.36
							59.50	6.90	21.34
<b>4i</b>	4-phenylpiperazino	C <sub>2</sub> H <sub>5</sub>	B	60	267 [a]	C <sub>18</sub> N <sub>2</sub> O <sub>4</sub>	66.65	6.22	17.27
							66.60	6.20	17.25
<b>4j</b>	4-ethoxycarbonyl-piperazino	C <sub>2</sub> H <sub>5</sub>	B	42	255 [c]	C <sub>15</sub> H <sub>20</sub> N <sub>4</sub> O <sub>4</sub>	56.24	6.29	17.49
							56.30	6.27	17.52

[a] From 2-ethoxyethanol. [b] From 1-Propanol. [c] From Ethanol/water.

Table 2  
Spectroscopic Data of Compounds 4

Compound No.	IR (nujol) $\nu$ $\text{cm}^{-1}$	$^1\text{H}$ NMR $\delta$ (ppm)
4a	3250, 2190, 1635, 1610, 1585	(DMSO- $d_6$ ): 1.80 (m, 4H, 2CH <sub>2</sub> ), 3.31 (s, 2H, CH <sub>2</sub> ), 3.54 (m 4H, CH <sub>2</sub> NCH <sub>2</sub> ), 5.20 (s, 1H, OH), 7.48 (s, 1H, H-3), 11.20 (br s, 1H, NH)
4b	3140, 2220, 1640, 1610, 1580	(DMSO- $d_6$ ): 3.34 (m 4H, CH <sub>2</sub> NCH <sub>2</sub> ), 3.61 (s, 2H, CH <sub>2</sub> ), 3.63 (m 4H, CH <sub>2</sub> OCH <sub>2</sub> ), 11.50 (br s, 1H, NH)
4c	3400, 2220, 1640, 1580	(DMSO- $d_6$ ): 2.35 (s, 3H, CH <sub>3</sub> ), 2.70, 3.53 (m, 8H piperaziny), 3.41 (s, 2H, CH <sub>2</sub> ), 10.55 (br s, 1H, NH)
4d	3400, 2200, 1640, 1600, 1580	(Pyridine- $d_5$ ): 3.21, 3.87 (m, 8H piperaziny), 3.77 (s, 2H, CH <sub>2</sub> ), 5.33 (br s, 2H, OH and NH), 6.85 (m, 3H, Ar), 7.20 (m, 2H, Ar)
4e	3490, 2220, 1700, 1650, 1590	(DMSO- $d_6$ ): 1.18 (t, 3H, CH <sub>3</sub> ), 3.40, 3.58 (m, 8H piperaziny), 3.60 (s, 2H, CH <sub>2</sub> ), 4.00 (q, 2H, CH <sub>2</sub> ), 11.20 (br s, 1H, NH)
4f	3240, 3140, 2210, 1640, 1590	(DMSO- $d_6$ ): 0.85 (t, 3H, CH <sub>3</sub> ), 1.82 (m, 4H, 2CH <sub>2</sub> ), 2.29 (q, 2H, CH <sub>2</sub> ), 3.52 (m 4H, CH <sub>2</sub> NCH <sub>2</sub> ), 7.40 (br s, 2H, OH and NH)
4g	3240, 2220, 1650, 1620	(DMSO- $d_6$ ): 0.88 (t, 3H, CH <sub>3</sub> ), 2.34 (q, 2H, CH <sub>2</sub> ), 3.38 (m 4H, CH <sub>2</sub> NCH <sub>2</sub> ), 3.61 (m 4H, CH <sub>2</sub> OCH <sub>2</sub> ), 10.40 (br s, 2H, OH and NH)
4h	3350, 2220, 1630, 1580	(Pyridine- $d_5$ ): 1.20 (t, 3H, CH <sub>3</sub> ), 2.02 (s, 3H, CH <sub>3</sub> ), 2.86 (q, 2H, CH <sub>2</sub> ), 2.32, 3.56 (m, 8H piperaziny), 5.36 (br s, 2H, OH and NH)
4i	3260, 2200, 1620, 1580	(Pyridine- $d_5$ ): 1.22 (t, 3H, CH <sub>3</sub> ), 2.87 (q, 2H, CH <sub>2</sub> ), 3.13 and 3.65 (m, 8H piperaziny), 5.25 (br s, 2H, OH and NH), 6.85 (m, 3H, Ar), 7.20 (m, 2H, Ar)
4j	3200, 2200, 1665, 1630, 1600, 1580	(Pyridine- $d_5$ ): 1.03 (t, 3H, CH <sub>3</sub> ), 1.20 (t, 3H, CH <sub>3</sub> ), 2.86 (q, 2H, CH <sub>2</sub> ), 3.47 (m, 8H piperaziny), 4.05 (q, 2H, CH <sub>2</sub> ), 5.60 (br s, 2H, OH and NH)

systems [4].

First the reactions between amidines **1** and carbon suboxide **2**, were carried out, which gave discrete yields of 6-amino-4-hydroxy-2(1*H*)-pyridone derivatives **4a-e**. The structure of the derivatives **4** was assigned on the basis of the analytical and spectroscopic data reported in Tables 1 and 2. Particularly,  $^1\text{H}$  nmr (DMSO- $d_6$ ) showed a singlet between 3.61 and 3.31 ppm related to the protons in C-3 and a broad signal between 11.50 and 10.55 ppm, which collapses after deuteration, due to the NH group. In most **4** derivatives, a further broad exchangeable singlet between 5.20 and 5.50 due to the OH group could be seen, which shows existence of a tautomeric equilibrium in solution, as had been seen in similar molecules synthesized by us [5].

At the same time, we reacted amidines **1** with the diethyl malonic acid without obtaining any result.

Subsequently using the bis-2,4,6-trichlorophenylmalonates **3a-b** as reagents, 4-hydroxy-2(1*H*)-pyridone derivatives were obtained in good yields. As expected, in solution C-3 substituted pyridones are to be found prevalently in the form of 4-hydroxy-2(1*H*)-pyridone, as can be seen from the  $^1\text{H}$  nmr spectra which clearly show the ethyl resonances at 0.85 ppm (triplet integrating for 3H with  $J = 7$ ) and 2.30 ppm (quartet, 2H) as well as broad bands for the NH and OH groups.

Carbon suboxide and active malonic esters proved useful in obtaining the 6-amino-4-hydroxy-2(1*H*)-pyridone derivatives in one step.

## EXPERIMENTAL

The melting points were determined on Kofler hot stage and are uncorrected. The ir spectra were obtained in nujol with a Perkin-Elmer 398 spectrophotometer. The  $^1\text{H}$  nmr spectra were recorded on a Varian Unity 300 spectrometer with shifts given in ppm downfield from internal hexamethyldisiloxane. The elemental analyses (C, H, N) were carried out with a Carlo Erba model 1106 Elemental Analyzer. The amidines **1** were obtained with a previously described procedure [4a].

6-Amino-4-hydroxy-2(1*H*)-pyridone Derivatives **4**. General Procedure.

Method A.

Carbon suboxide (16 mmoles) was added during one hour at  $-70^\circ$  to a stirred solution of amidine **1** (16 mmoles) in dry dichloromethane. When the addition was completed, the mixture was stirred at  $0^\circ$  for 5 hours, and then kept at room temperature with stirring for 48 hours. The precipitate was filtered and crystallized from a suitable solvent to give pyridones **4**.

Method B.

A mixture of amidine **1** (10 mmoles) and bis-2,4,6-trichlorophenylmalonate **3** (10 mmoles) was heated to  $150-160^\circ$  for 5 hours. After cooling, the residue was treated with ethyl acetate and the resulting precipitate was filtered, washed with ethyl acetate and recrystallized.

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